# Metadata

#### **Dataset Title:**

Cary\_Stream\_Chemistry\_Data.csv Cary Environmental Monitoring Program Stream Chemistry Data: 1985-2019

## Abstract

The Cary Institute of Ecosystem Studies Environmental Monitoring Program is a long-term data collection program designed to understand how the environment changes over time. The program includes monitoring of climate including temperature and precipitation, as well as variables related to air pollution, such as acid deposition and ozone, water pollution and streamwater hydrology. The Cary Institute of Ecosystem Studies, Environmental Monitoring Program furnishes data under the following conditions: The data have received quality assurance scrutiny by our program, and, although we are confident of the accuracy of these data, the Cary Institute will not be held liable for errors in these data. Data are subject to change resulting from updates in data screening or models used. Data citation: The following is a standard citation for referencing data from the Cary Institute of Ecosystem Studies, Environmental Monitoring Program Institute of Ecosystem Studies, Environmental Monitoring Program and program and although we are confident of the accuracy of these data, the Cary Institute will not be held liable for errors in these data.

Cary Institute of Ecosystem Studies, Environmental Monitoring Program. 2008 (or current year). Cary Institute of Ecosystem Studies, Box AB, Millbrook, NY 12545, www.caryinstitute.org.

Those wishing to publish data from the Cary Institute of Ecosystem Studies, Environmental Monitoring Program are encouraged to contact Data Manager Vicky Kelly, kellyv@caryinstitute.org.

## Investigators

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# Keywords

Cary Institute, Cary Institute of Ecosystem Studies, data, climate, stream, streamwater, chemistry, conductivity, conductance, pH, calcium, magnesium, potassium, sodium, chloride, sulfate, nitrate, ammonium, phosphate, silica, dissolved organic carbon

# Timeframe

The data in this file start 17 January 1985 and end 28 December 2019. Data collection is ongoing. Sample collection began in November 1985 (Site 2) and January 1985 (Site 4).

## **Geographic location**

Stream samples are collected at the end of every month at two sites on the East Branch of Wappinger Creek at the Cary Institute of Ecosystem Studies. Before 1996, samples were collected at four sites. Redundancy in the data and cost of analysis prompted us to discontinue collecting samples at two of the sites. The two remaining sites are Site 2 (Fern Glen, GPS coordinates N41.78707, W73.73317) and Site 4 (Lowlands, GPS coordinates N41.79235, W73.74937).

## **Methods**

The stream is a tributary to the main branch of Wappinger Creek, which flows into the Hudson River at

Wappingers Falls. Approximately 1.6 km upstream from the gauging station is the Village of Millbrook sewage treatment plant. In addition to the sewage treatment plant, Dietrich Pond and its associated dam can influence the stream. Stream samples are collected at the end of every month. Samples are collected when the stream is as near base flow as possible, however, base flow varies seasonally and is generally higher in winter than summer. The samples are analyzed at the Cary Institute Analytical Laboratory for pH, conductivity, nitrate, sulfate, ammonium, phosphate, chloride, sodium, calcium, magnesium, potassium and silica and DOC (see Table below for analytical methods). Samples collected between 1985 and 1998 were not preserved in any way and were stored in the light at room temperature before analyses were completed. Samples from January 1999 onward are preserved as follows. Samples are divided into three aliquots, one aliquot is filtered and preserved with Optima grade concentrated sulfuric acid (0.5 ml per 125 ml of sample) for analysis of nitrate, ammonium, phosphate, and DOC. Another aliquot is filtered and analyzed for sulfate, chloride, sodium, calcium, magnesium, potassium and silica. Silica analyses are completed within 4 days of sample collection. The third aliquot is untreated and analyzed for conductivity and pH.

#### STREAM CHEMISTRY DATA QUALITY ASSURANCE & QUALITY CONTROL PARAMETERS & METHODS

All sample bottles and glassware are cleaned with deionized and air dried. Once each year a sample bottle is filled with deionized water, filtered, acidified and analyzed as a regular sample. This is to ensure that the sample handling procedures introduce no contamination.

When analytical results are received from the Cary Institute analytical lab, data are checked by examining time series graphs of sample concentrations for each analyte. Any data that are potential outliers are reanalyzed. If reanalysis returns the same results, no changes are made to the dataset. If reanalysis returns different results, ion balances are examined and the sample is examined for potential contamination. The concentration value that returns the best ion balance is submitted to the dataset unless there is obvious contamination, in which case the data value is replaced with a missing value code.

Column name	Description	Unit or code explanation or date format	Empty value code
	Date in MM/DD/YYYY format where MM is month number, DD is date		
DATE	number and YYYY is year		blank
SITE	Site number on Wappinger Creek	Unitless	blank
COND	Conductivity (umho)	Umho	blank
СА	Concentration of Ca+2 (mg/L) (DL 0.01 mg/L)	mg/L	blank
MG	Concentration of Mg+2 (mg/L) (DL 0.01 mg/L)	mg/L	blank
к	Concentration of K+ (mg/L) (DL 0.01 mg/L)	mg/L	blank
NA	Concentration of Na+ (mg/L) (DL 0.01 mg/L)	mg/L	blank
NH4_N	Concentration of NH4+_N (mg/L) (DL 0.02 mg/L)	mg/L	blank
РН	рН	pH units	blank

#### **Data Table**

SO4	Concentration of SO4-2 (mg/L) (DL 0.02 mg/L)	mg/L	blank
NO3_N	Concentration of NO3N (mg/L) (DL 0.02 mg/L)	mg/L	blank
CL	Concentration of Cl- (mg/L) (DL 0.02 mg/L)	mg/L	blank
PO4_P	Concentration of PO4-3_P (mg/L) (DL 0.002 mg/L)	mg/L	blank
SIO2	Concentration of SiO2 (mg/L) (DL 0.1 mg/L)	mg/L	blank
DOCPPM	Concentration of DOC (ppm) (DL 0.05 ppm)	Ppm	blank
ТЕМР	Stream temperature at time of sampling (C) (Site 2 only)	degrees C	blank
Q	Stream flow rate at time sample collected (m^3/s) (Site 2 only)	m^3/s	blank

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#### Methods of Analysis Cary Institute of Ecosystem Studies Analytical Laboratory

ION	INSTRUMENT	TECHNIQUE
NH4 <sup>+</sup>	Lachat QuikChem 8000	Phenate method <sup>1</sup> #10-107-06-1-J
SO₄⁼, NO₃⁻ (PPT, AQ), Cl⁻	Dionex ICS2000 Ion Chromatograph	Ion exchange chromatography, AS18 and AG18 columns, SRS (self-regenerating) suppressor <sup>2</sup> with CRD 200 (carbonate removal device)
K+, Na+	Perkin Elmer Aanalyst 300 Atomic Absorption Spectrometer	Flame atomization, direct air <sup>3</sup>
Ca <sup>++</sup> , Mg <sup>++</sup>	Leeman Labs Inductively Coupled Plasma/Profile	Emission spectroscopy
NO <sub>3</sub> - (WC)	Lachat QuikChem 8000	Cadmium diazotization <sup>1</sup> Method #10-107-04-1-C
PO <sub>4</sub>	Lachat QuikChem 8000	Phosphomolybdate <sup>1</sup> Method #_10-115-01-1-M
рН	Fisher-Accumet AR20 pH meter with Fisher glass electrode, Fisher calomel reference probe	Standardization with Fisher 7.00 and 3.00 buffer solutions; samples and buffers at room
Specific Conductance	Fisher-Accumet AR20 pH/conductivity meter	temperature Conductivity probe w/ 1.0 cm <sup>-1</sup> cell constant
DOC (Dissolved Organic Carbon)	Shimadzu TOC 5050	High temperature combustion of sample; platinum catalyst C to CO2, NDIR detect.

<sup>&</sup>lt;sup>1</sup>Standard Lachat methods, 2000, Lachat Instruments, Milwaukee, WI

<sup>&</sup>lt;sup>2</sup> Small, H., Stevens, T.S., and Bauman, W.C. Anal. Chem. 1975, 47:1801-1809

<sup>&</sup>lt;sup>3</sup> Slavin, W. Atomic absorption spectroscopy. Wiley-Interscience, New York. 1968. PPT=precipitation

samples, AQ=air samples, WC=Wappinger Creek samples